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meso-3,6-Dioxopiperazine-2,5-diacetamide

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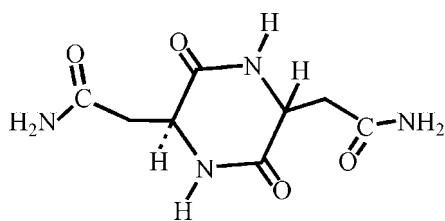
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Key indicators: single-crystal X-ray study; $T = 293$ K; mean $\sigma(\text{C}-\text{C}) = 0.002$ Å; R factor = 0.037; wR factor = 0.098; data-to-parameter ratio = 16.0.

The title compound, $\text{C}_8\text{H}_{12}\text{N}_4\text{O}_4$, was obtained by cyclization of the two L-asparagine molecules and reveals a crystallographic inversion symmetry, and accordingly the two stereogenic centres are of opposite chirality. Thus, an asymmetric unit comprises a half of a molecule. The molecules are assembled into a three-dimensional hydrogen-bonding network by $\text{N}-\text{H}\cdots\text{O}$ hydrogen bonds.

Related literature

For general background to coordination polymers, see: Anitha *et al.* (2005); Aarthy *et al.* (2005); Guenifa *et al.* (2009); Moussa Slimane *et al.* (2009). For related structures, see: Howes *et al.* (1983).



Experimental

Crystal data

 $\text{C}_8\text{H}_{12}\text{N}_4\text{O}_4$ $M_r = 228.22$ Monoclinic, $P2_1/c$ $a = 5.0409$ (10) Å $b = 8.3178$ (17) Å $c = 12.900$ (3) Å $\beta = 109.76$ (3)° $V = 509.0$ (2) Å³ $Z = 2$ Mo $K\alpha$ radiation $\mu = 0.12$ mm⁻¹
 $T = 293$ K

0.10 × 0.10 × 0.10 mm

Data collection

Rigaku R-Axis RAPID
diffractometer

Absorption correction: multi-scan

(ABSCOR; Higashi, 1995)

 $T_{\min} = 0.988$, $T_{\max} = 0.988$

4836 measured reflections

1166 independent reflections

889 reflections with $I > 2\sigma(I)$ $R_{\text{int}} = 0.028$

Refinement

 $R[F^2 > 2\sigma(F^2)] = 0.037$ $wR(F^2) = 0.098$ $S = 1.07$

1166 reflections

73 parameters

H-atom parameters constrained

 $\Delta\rho_{\text{max}} = 0.24$ e Å⁻³ $\Delta\rho_{\text{min}} = -0.16$ e Å⁻³

Table 1

Hydrogen-bond geometry (Å, °).

$D-\text{H}\cdots A$	$D-\text{H}$	$\text{H}\cdots A$	$D\cdots A$	$D-\text{H}\cdots A$
$\text{N1}-\text{H1A}\cdots\text{O2}^{\text{ii}}$	0.86	2.12	2.9185 (19)	154
$\text{N1}-\text{H1B}\cdots\text{O2}^{\text{ii}}$	0.86	2.03	2.8795 (18)	167
$\text{N2}-\text{H2C}\cdots\text{O1}^{\text{iii}}$	0.86	2.06	2.8509 (17)	152

Symmetry codes: (i) $-x, y - \frac{1}{2}, -z + \frac{1}{2}$; (ii) $x, -y + \frac{1}{2}, z - \frac{1}{2}$; (iii) $x + 1, y, z$.

Data collection: *RAPID-AUTO* (Rigaku, 1998); cell refinement: *RAPID-AUTO*; data reduction: *CrystalStructure* (Rigaku/MS, 2004); program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *SHELXTL* (Sheldrick, 2008); software used to prepare material for publication: *SHELXL97*.

This project was supported by the Scientific Research Fund of Zhejiang Provincial Education Department (grant No. Y201017782). Thanks are also extended to the K. C. Wong Magna Fund of Ningbo University.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: KP2350).

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supplementary materials

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meso-3,6-Dioxopiperazine-2,5-diacetamide

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Comment

The past decade has witnessed enormous expansion of research on non-centrosymmetric coordination polymers. For such purpose, rational design and synthesis have been focused on choices of metal cations with non-centrosymmetric organic ligands. Asparagine (Anitha *et al.*, (2005); Aarthy *et al.*, (2005); Guenifa *et al.*, (2009); Moussa Slimane *et al.*, (2009)) is a chiral molecule and one of the common neutral amino acids with carboxamide as the side-chain functional group. However, condensation led to a centrosymmetric compound and we report its crystal structure.

In (I) (Fig. 1), two *L*-asparagine molecules engage in the dehydration condensation between each carboxyl and the adjacent amino groups. The resulting product reveals the molecular symmetry C_i (crystallographic inversion symmetry). In (I) a piperazinedione-2,5 unit is close to be planar (the mean value of intracyclic torsion angles is 2.65°) and it is different to those reported by (Howes *et al.*, (1983)). The molecules are connected through $N1-H1A\cdots O2^i$, $N1-H1B\cdots O2^{ii}$, and $N2-H2C\cdots O1^{iii}$ hydrogen bonds generating a 3D-network (Table 1, Figs. 2 and 3).

Experimental

Dropwise addition of 1 *M* NaOH (1.0 mL) to a stirred aqueous solution of (0.1438 g, 0.5 mmol) $ZnSO_4 \cdot 7H_2O$ in 5.0 mL H_2O produced pale-white $Zn(OH)_2 \cdot xH_2O$ precipitate, which was separated by centrifugation and washed with distilled water for several times. Subsequently, the 0.1501 g (1.0 mmol) *L*-asparagine was dissolved completely with 10.0 mL H_2O , and then the precipitate was added. The resulting mixture was further stirred at 323 K for 1 h and then filtered. The white filtrate was allowed to stand at room temperature. Slow evaporation for several days afforded colourless needle-like crystals.

Refinement

H atoms bonded to C atoms were placed in their geometrically calculated positions and refined using the riding model, with C–H distances 0.93 Å and $U_{iso}(H) = 1.2 U_{eq}(C)$.

Figures

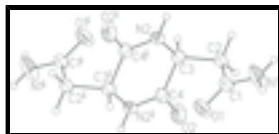


Fig. 1. ORTEP view of the title compound. The displacement ellipsoids are drawn at 45% probability displacement ellipsoids. [Symmetry codes: (i) $-x+1, -y, -z+1$.]

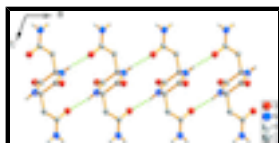


Fig. 2. Packing diagram of the title crystal structure viewed down along [010] direction with $N2-H2C\cdots O1$ hydrogen bond motif.

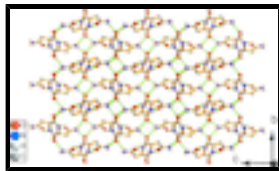


Fig. 3. Packing diagram of the title crystal viewed down the a axis shows 3D-hydrogen bond network. N–H···O hydrogen bonds are shown as dashed lines.

3,6-Dioxopiperazine-2,5-diacetamide

Crystal data

$C_8H_{12}N_4O_4$

$M_r = 228.22$

Monoclinic, $P2_1/c$

Hall symbol: $-P\ 2ybc$

$a = 5.0409\ (10)\ \text{\AA}$

$b = 8.3178\ (17)\ \text{\AA}$

$c = 12.900\ (3)\ \text{\AA}$

$\beta = 109.76\ (3)^\circ$

$V = 509.0\ (2)\ \text{\AA}^3$

$Z = 2$

$F(000) = 240$

$D_x = 1.489\ \text{Mg m}^{-3}$

Mo $K\alpha$ radiation, $\lambda = 0.71073\ \text{\AA}$

Cell parameters from 3368 reflections

$\theta = 3.4\text{--}27.4^\circ$

$\mu = 0.12\ \text{mm}^{-1}$

$T = 293\ \text{K}$

Needle, colourless

$0.10 \times 0.10 \times 0.10\ \text{mm}$

Data collection

Rigaku R-Axis RAPID
diffractometer

Radiation source: fine-focus sealed tube
graphite

Detector resolution: 0 pixels mm^{-1}

ω scans

Absorption correction: multi-scan
(*ABSCOR*; Higashi, 1995)

$T_{\min} = 0.988$, $T_{\max} = 0.988$

4836 measured reflections

1166 independent reflections

889 reflections with $I > 2\sigma(I)$

$R_{\text{int}} = 0.028$

$\theta_{\max} = 27.5^\circ$, $\theta_{\min} = 3.4^\circ$

$h = -6 \rightarrow 5$

$k = -10 \rightarrow 10$

$l = -16 \rightarrow 16$

Refinement

Refinement on F^2

Least-squares matrix: full

$R[F^2 > 2\sigma(F^2)] = 0.037$

$wR(F^2) = 0.098$

$S = 1.07$

1166 reflections

73 parameters

0 restraints

Primary atom site location: structure-invariant direct methods

Secondary atom site location: difference Fourier map

Hydrogen site location: inferred from neighbouring sites

H-atom parameters constrained

$w = 1/[\sigma^2(F_o^2) + (0.0415P)^2 + 0.146P]$

where $P = (F_o^2 + 2F_c^2)/3$

$(\Delta/\sigma)_{\max} < 0.001$

$\Delta\rho_{\max} = 0.24\ \text{e \AA}^{-3}$

$\Delta\rho_{\min} = -0.16\ \text{e \AA}^{-3}$

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R -factor wR and goodness of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating R -factors(gt) *etc.* and is not relevant to the choice of reflections for refinement. R -factors based on F^2 are statistically about twice as large as those based on F , and R -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
O1	0.0282 (2)	-0.01262 (16)	0.31172 (9)	0.0483 (4)
O2	0.2805 (3)	0.29340 (13)	0.49666 (9)	0.0482 (3)
N1	0.0676 (4)	0.0393 (2)	0.14760 (11)	0.0580 (5)
H1A	-0.0736	-0.0213	0.1141	0.070*
H1B	0.1561	0.0889	0.1107	0.070*
N2	0.6588 (2)	-0.04206 (14)	0.43619 (9)	0.0316 (3)
H2C	0.7546	-0.0678	0.3948	0.038*
C1	0.1482 (3)	0.05588 (18)	0.25545 (11)	0.0320 (3)
C2	0.3991 (3)	0.16430 (17)	0.30689 (11)	0.0300 (3)
H2A	0.3353	0.2749	0.3022	0.036*
H2B	0.5274	0.1554	0.2656	0.036*
C3	0.5568 (3)	0.12212 (17)	0.42763 (11)	0.0289 (3)
H3A	0.7224	0.1923	0.4531	0.035*
C4	0.3792 (3)	0.15646 (18)	0.49910 (11)	0.0307 (3)

Atomic displacement parameters (\AA^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
O1	0.0425 (6)	0.0752 (9)	0.0326 (6)	-0.0194 (6)	0.0199 (5)	-0.0009 (5)
O2	0.0730 (8)	0.0412 (6)	0.0366 (6)	0.0250 (6)	0.0267 (6)	0.0066 (5)
N1	0.0743 (11)	0.0739 (11)	0.0270 (7)	-0.0402 (9)	0.0185 (7)	-0.0063 (7)
N2	0.0336 (6)	0.0399 (7)	0.0266 (6)	0.0094 (5)	0.0172 (5)	0.0028 (5)
C1	0.0331 (7)	0.0392 (8)	0.0267 (7)	0.0000 (6)	0.0140 (6)	0.0016 (6)
C2	0.0347 (7)	0.0327 (7)	0.0261 (7)	-0.0005 (6)	0.0150 (6)	0.0024 (6)
C3	0.0292 (7)	0.0322 (7)	0.0268 (7)	-0.0005 (6)	0.0116 (6)	-0.0003 (6)
C4	0.0329 (7)	0.0364 (7)	0.0227 (7)	0.0067 (6)	0.0093 (6)	0.0003 (6)

Geometric parameters (\AA , $^\circ$)

O1—C1	1.2304 (17)	C1—C2	1.512 (2)
O2—C4	1.2392 (17)	C2—C3	1.5309 (19)
N1—C1	1.3182 (19)	C2—H2A	0.9700
N1—H1A	0.8599	C2—H2B	0.9700

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N1—H1B	0.8599	C3—C4	1.5135 (19)
N2—C4 ⁱ	1.3219 (18)	C3—H3A	0.9800
N2—C3	1.4502 (18)	C4—N2 ⁱ	1.3219 (18)
N2—H2C	0.8599		
C1—N1—H1A	119.9	C1—C2—H2B	109.1
C1—N1—H1B	120.1	C3—C2—H2B	109.1
H1A—N1—H1B	120.0	H2A—C2—H2B	107.9
C4 ⁱ —N2—C3	127.05 (12)	N2—C3—C4	113.51 (11)
C4 ⁱ —N2—H2C	116.4	N2—C3—C2	110.01 (11)
C3—N2—H2C	116.5	C4—C3—C2	111.46 (11)
O1—C1—N1	122.47 (14)	N2—C3—H3A	107.2
O1—C1—C2	121.49 (13)	C4—C3—H3A	107.2
N1—C1—C2	116.05 (13)	C2—C3—H3A	107.2
C1—C2—C3	112.30 (12)	O2—C4—N2 ⁱ	122.34 (13)
C1—C2—H2A	109.1	O2—C4—C3	118.24 (13)
C3—C2—H2A	109.1	N2 ⁱ —C4—C3	119.39 (12)

Symmetry codes: (i) $-x+1, -y, -z+1$.

Hydrogen-bond geometry ($\text{\AA}, ^\circ$)

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
N1—H1A \cdots O2 ⁱⁱ	0.86	2.12	2.9185 (19)	154.
N1—H1B \cdots O2 ⁱⁱⁱ	0.86	2.03	2.8795 (18)	167.
N2—H2C \cdots O1 ^{iv}	0.86	2.06	2.8509 (17)	152.

Symmetry codes: (ii) $-x, y-1/2, -z+1/2$; (iii) $x, -y+1/2, z-1/2$; (iv) $x+1, y, z$.

Fig. 1

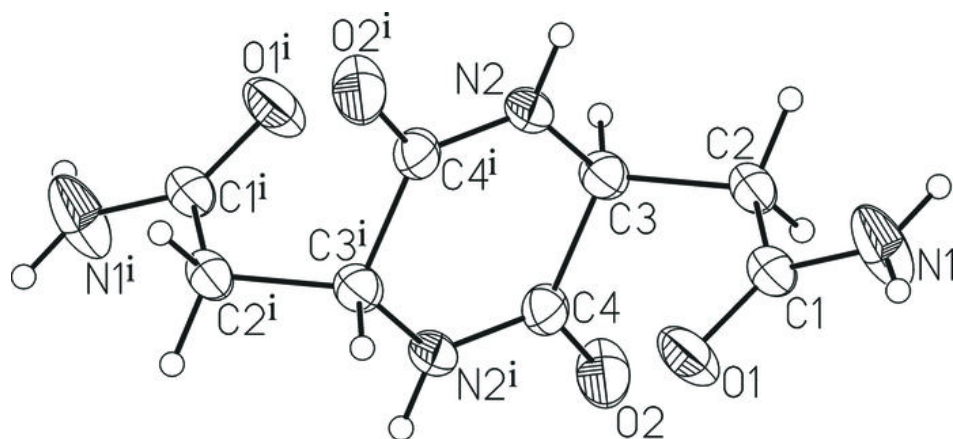


Fig. 2

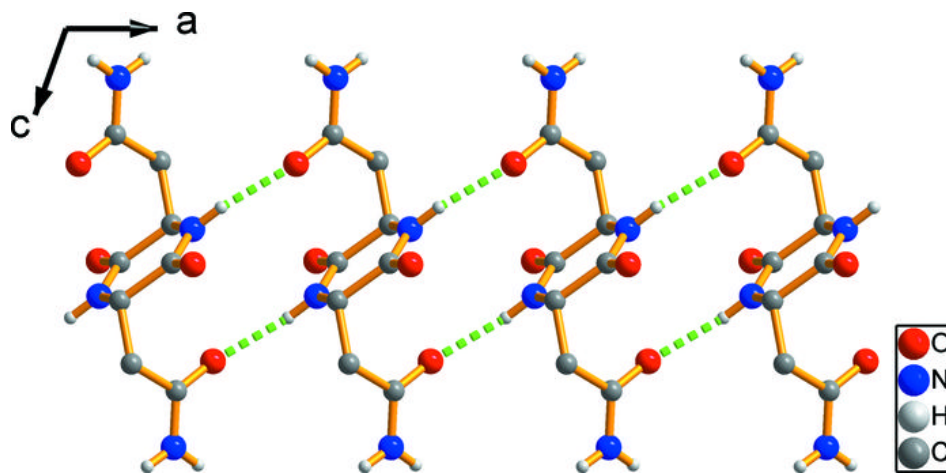


Fig. 3

